# ORGANIC CHEMISTRY ALKENE REACTIONS: HALOGENATIONS AND HYDRATION 

## Alkene halogenation ( $\mathrm{X}_{2}$ )

This reaction requires an alkene and a halogen $X_{2}\left(\mathrm{Br}_{2}, \mathrm{I}_{2}, \mathrm{~F}_{2}\right)$.

Reaction: add two halogen atoms $(\mathrm{X})$ to the carbon atoms on a double bond.


## PRACTICE PROBLEM 1

Complete each reaction by drawing the correct missing reactant or product.





## PRACTICE PROBLEM 1

Complete each reaction by drawing the correct missing reactant or product.

- answer -


$+$


$+$


$\mathrm{Cl}_{2}$



## Alkene hydrohalogenation (HX)

This reaction requires an alkene and an $\mathrm{HX}(\mathrm{X}=\mathrm{Br}, \mathrm{Cl}, \mathrm{I})$ molecule.

Reaction: add a hydrogen $(H)$ atom and a halogen $(X)$ atom across a double bond
Product: the halogen $(\mathrm{X})$ atom adds to the more substituted carbon atom, and the hydrogen $(\mathrm{H})$ atoms adds to the less substituted carbon atom. This is called Markovnikov's Rule.


## PRACTICE PROBLEM 2

Complete each reaction by drawing the correct missing reactant or product.







## PRACTICE PROBLEM 2

Complete each reaction by drawing the correct missing reactant or product.


$+$




## Alkene hydration ( $\mathrm{H}_{2} \mathrm{O}, \mathrm{H}^{+}$)

This reaction requires an alkene, a water molecule ( $\mathrm{H}_{2} \mathrm{O}$ ), and an acid catalyst ( $\mathrm{H}^{+}, \mathrm{H}_{3} \mathrm{O}^{+}, \mathrm{H}_{2} \mathrm{SO}_{4}$, etc.).

Reaction: add a hydrogen ( H ) atom and a hydroxide $(-\mathrm{OH})$ group across a double bond
Product: the hydroxide $(-\mathrm{OH})$ group adds to the more substituted carbon atom, and the hydrogen $(\mathrm{H})$ atoms adds to the less substituted carbon atom. This still follows Markovnikov's Rule.


2-methylprop-2-ene
$\left(\mathrm{C}_{5} \mathrm{H}_{10}\right)$

2-methylbutan-2-ol

$$
\left(\mathrm{C}_{5} \mathrm{H}_{11} \mathrm{OH}\right)
$$

## PRACTICE PROBLEM 3

Complete each reaction by drawing the correct missing reactant or product.






## PRACTICE PROBLEM 3

Complete each reaction by drawing the correct missing reactant or product.

\author{

- answer -
}






## PRACTICE PROBLEM 4

Which of the following reaction(s) would yield products with a new chiral center?

## - answer -




$+\mathrm{H}-\mathrm{Br} \longrightarrow$

$+\mathrm{Br}-\mathrm{Br} \longrightarrow$

$+\mathrm{H}-\mathrm{Cl} \longrightarrow$

## PRACTICE PROBLEM 4

Which of the following reaction(s) would yield products with a new chiral center?

## - ans\%er -






