



# BIOCHEMISTRY

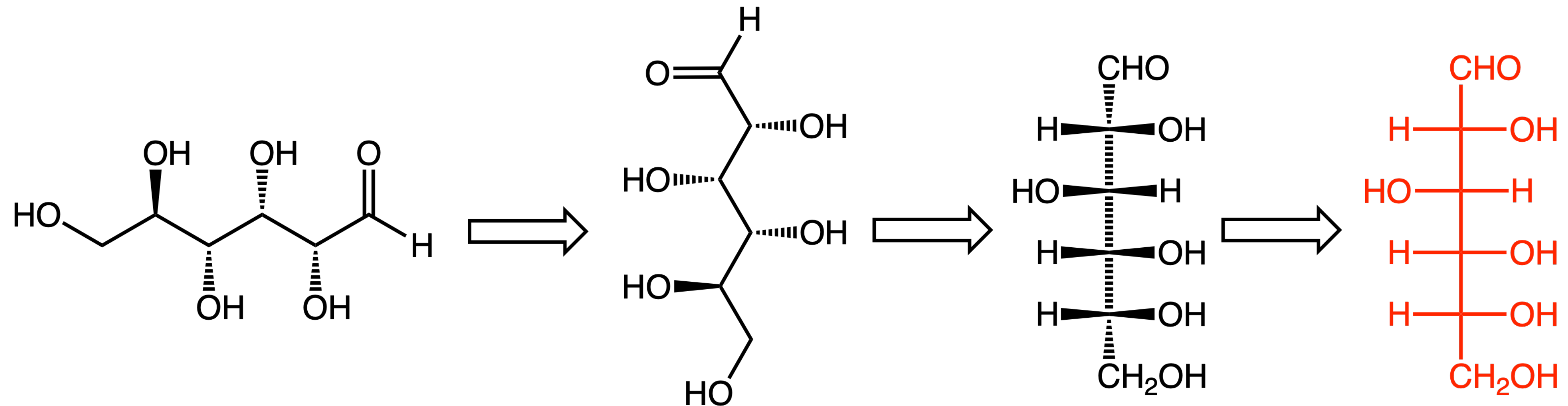
SUGARS: FISCHER & HAWORTH PROJECTIONS

CHEMISTRY 165 // SPRING 2020

# What is a sugar?

Most sugars have the general formula:  $C_x(H_2O)_y$ .

A single sugar unit (the simplest carbohydrate) is often referred to as a monosaccharide. When a polymer of monosaccharides is formed, we call it a polysaccharide (complex carbohydrate).



Because sugars have many chiral centers, they are often drawn using the **Fischer Projection**, which is shown on the far right-hand side.

# Fischer projections: What do they tell us?

The **Fischer projection** of a sugar is a standardized shorthand for representing sugars.

Let's take the example of glucose, which is shown to the right in two forms/isomers.

Contained in the Fischer projection are several important pieces of information:

- The number of carbon atoms in the sugar (-ose).

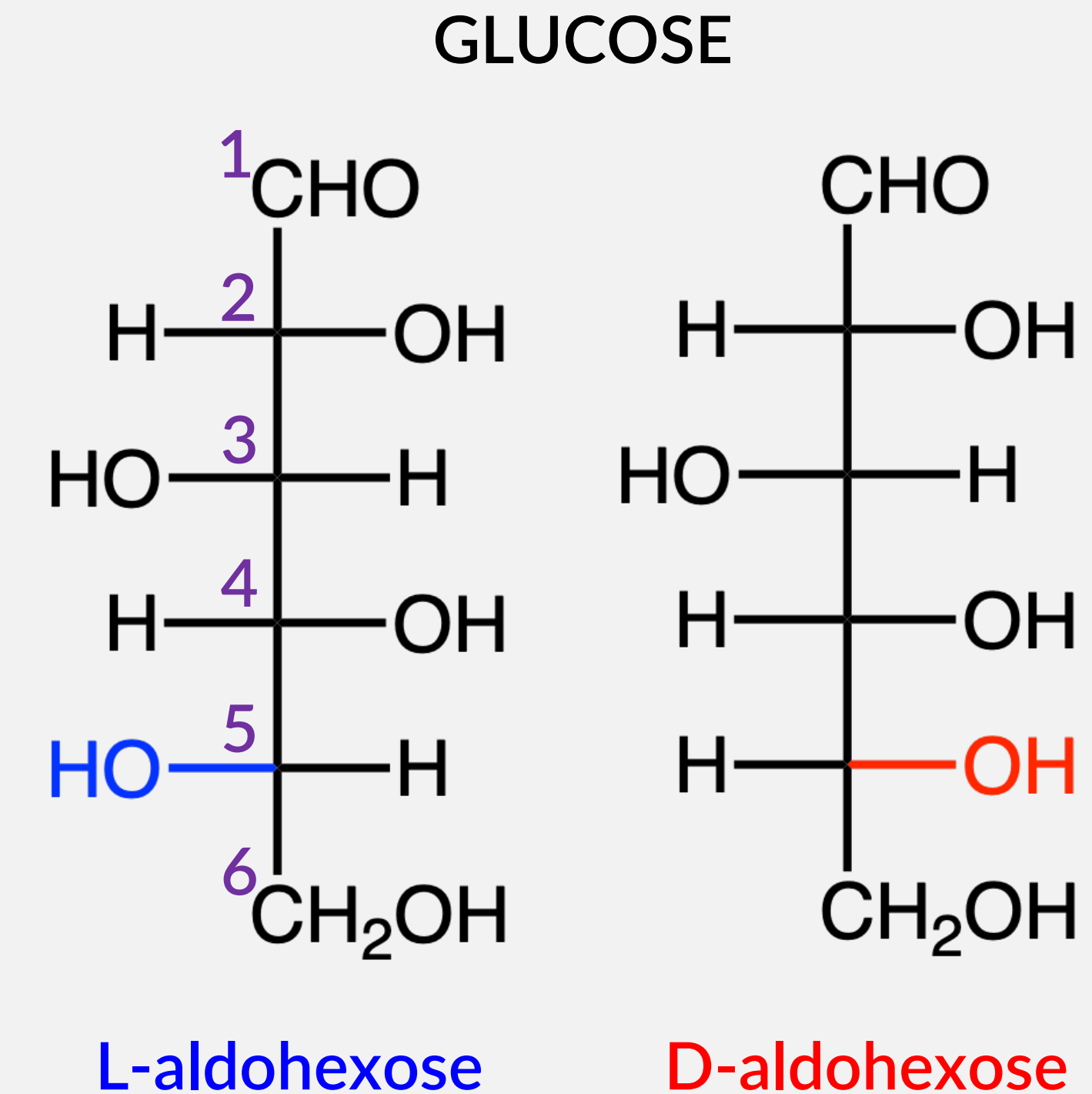
*Glucose is a 6-carbon sugar, so it is a hexose.*

- The placement of the carbonyl (C=O) group, and whether it is a ketone (ketose) or aldehyde (aldose). The C=O is always placed at the beginning (top) of the sugar/saccharide.

*Glucose is an aldehyde, so it is an aldohexose.*

- The configuration or stereochemistry of the last stereocenter (or chiral center).

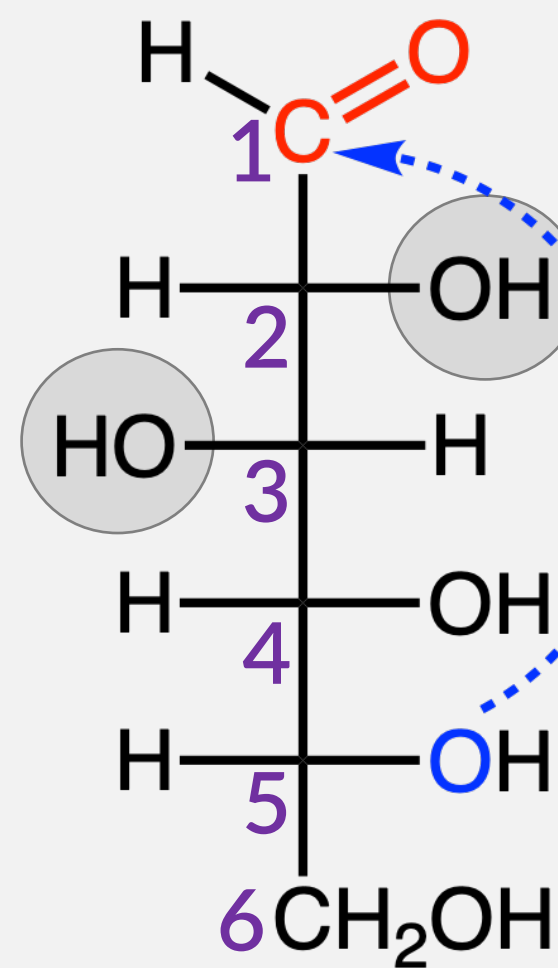
- If the **-OH** is on the **left** side, it is an L-sugar.
- If the **-OH** is on the **right** side, it is a D-sugar.



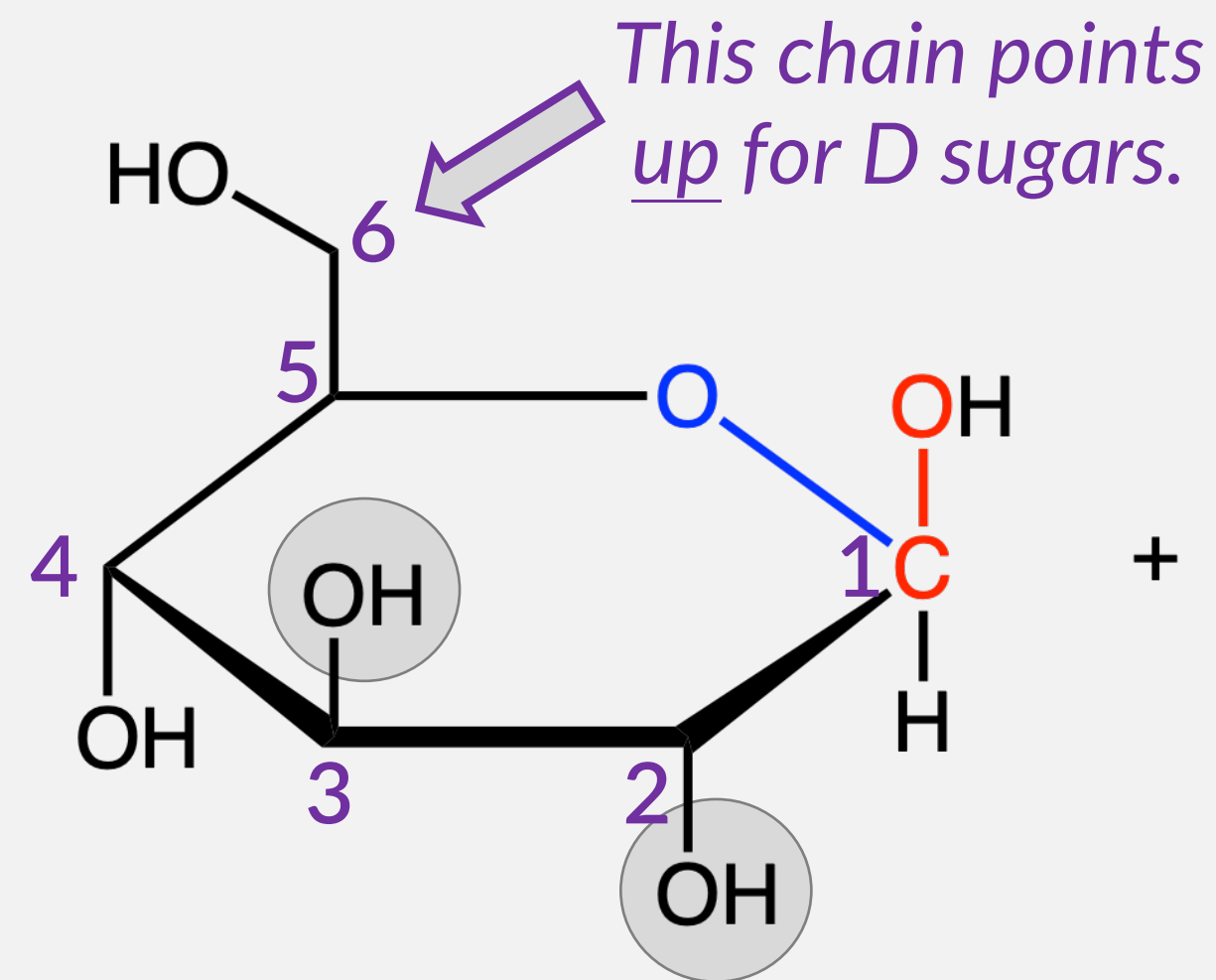
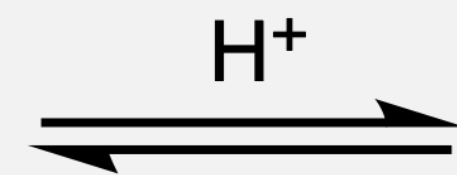
# Haworth projections: Cyclic sugars (Ex. 1)

Sugars often cyclize into 5- (furanose) or 6-membered (pyranose) rings in of acidic ( $H^+$ ) medium.

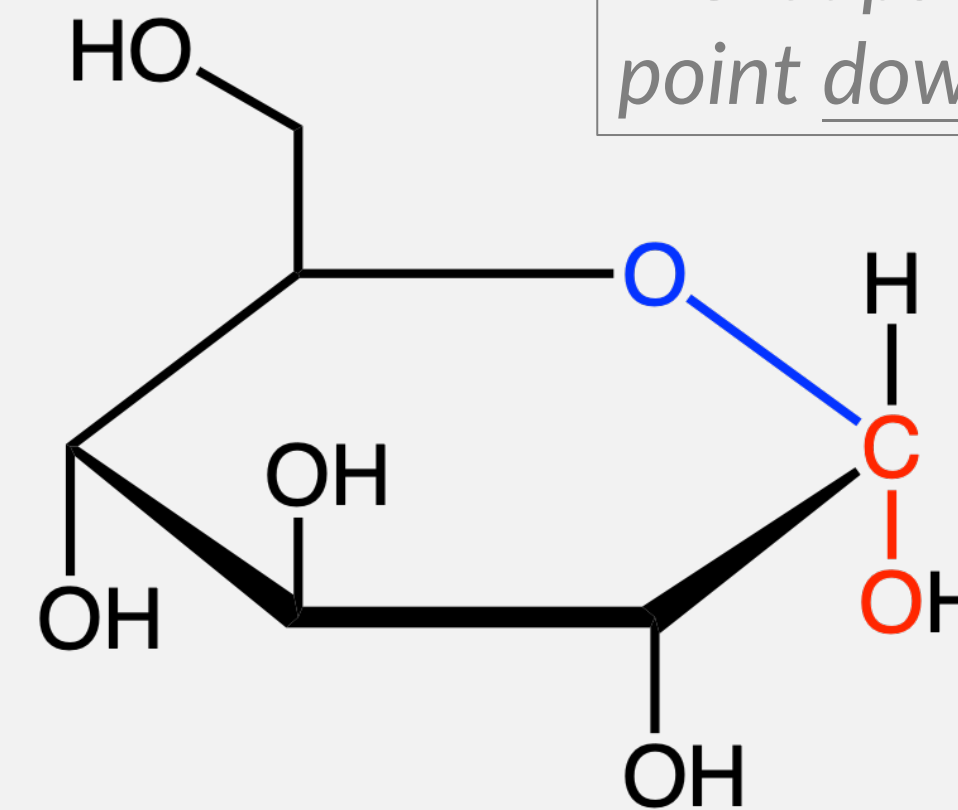
Let's take the example of D-glucose (a D-aldohexose).



**D-glucose**  
(aldohexose)



**$\beta$ -D-glucopyranose**  
( $\beta$   $\rightarrow$  -OH on  $^1C$  points up)



**$\alpha$ -D-glucopyranose**  
( $\alpha$   $\rightarrow$  -OH on  $^1C$  points down)

Groups on the left point up on the ring.  
Groups on the right point down on the ring.

The anomeric carbon is the  $^1C$  which is the only carbon attached to two O atoms.

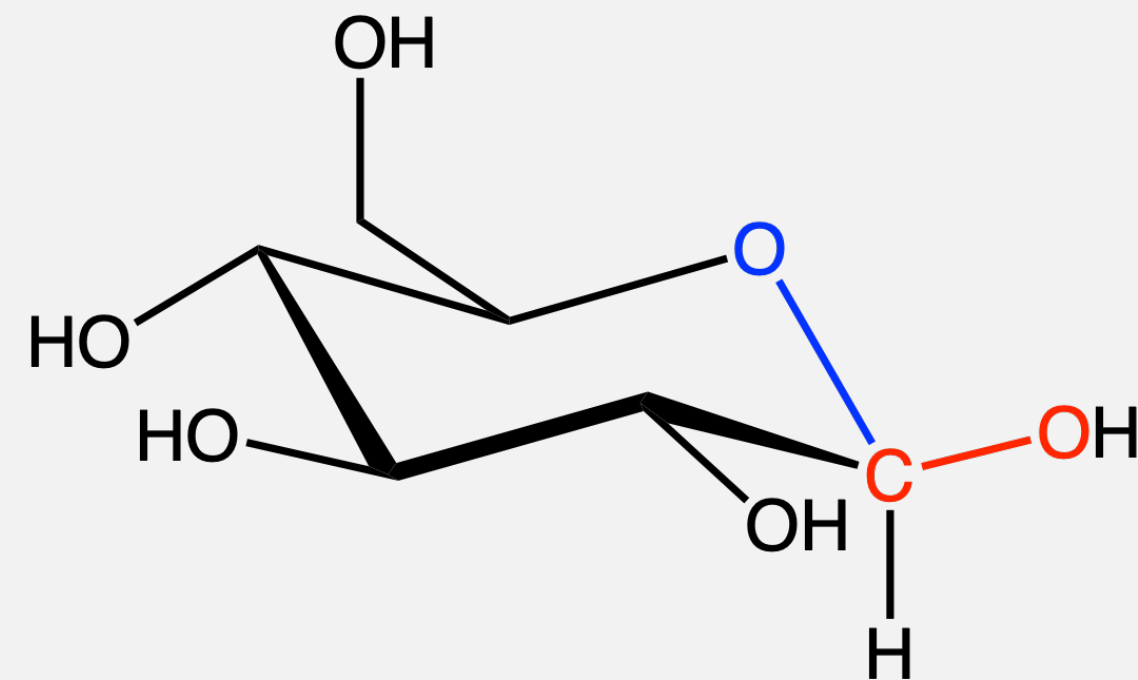
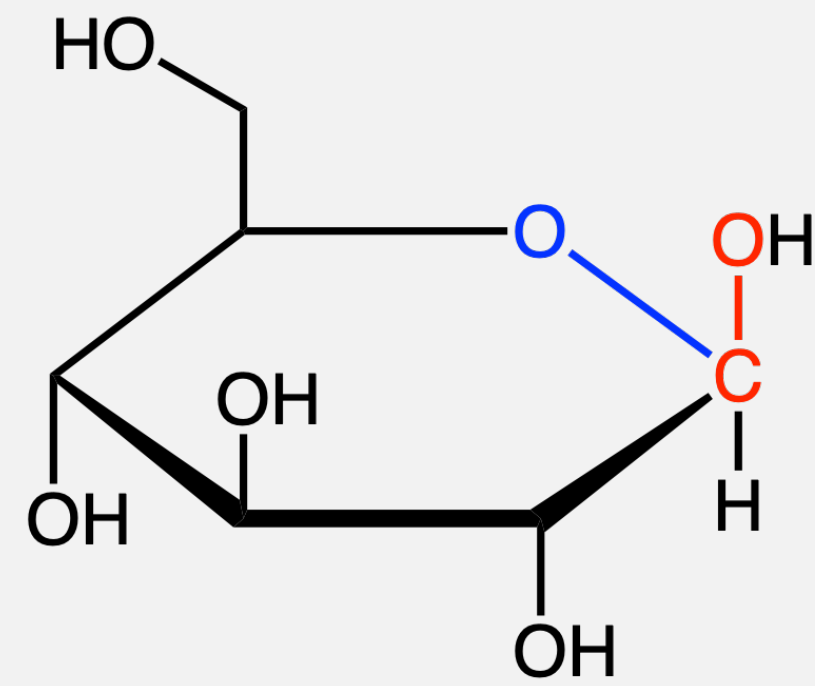
The  $\alpha$ - and  $\beta$ -pyranoses are isomers that are called anomers.

# Chair conformations

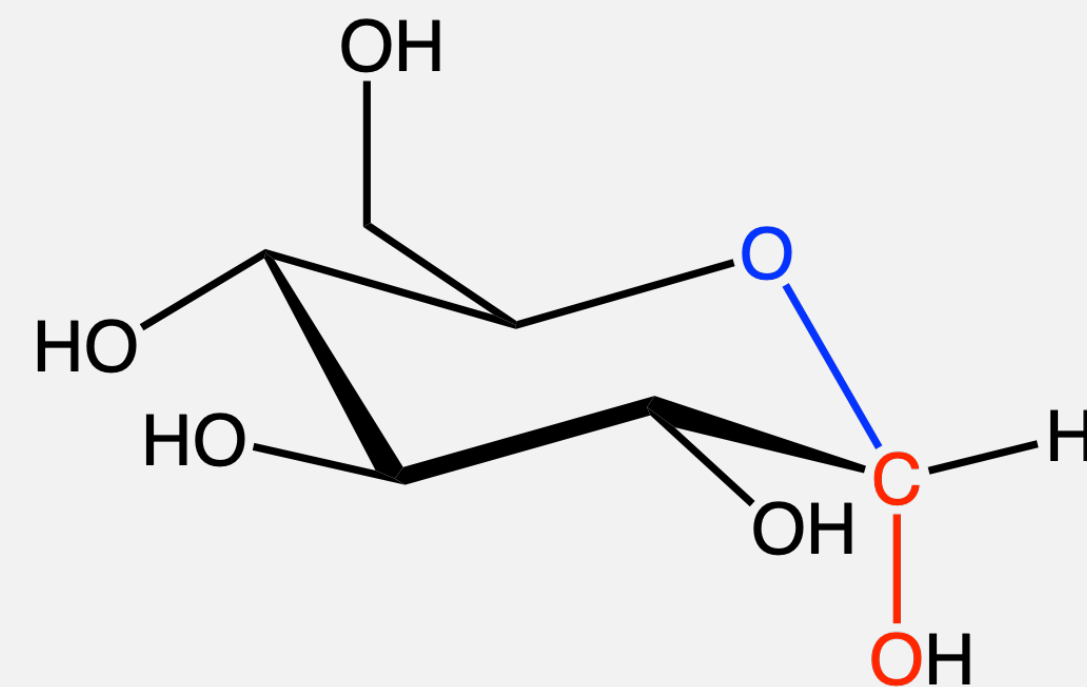
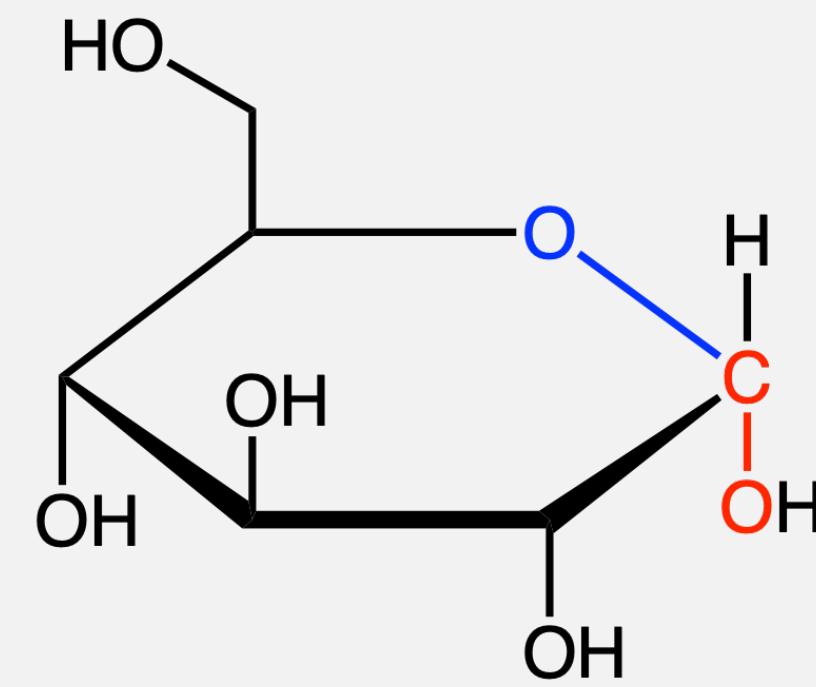
6-membered (pyranose) rings exist in a chair conformation because of the angles on the  $sp^3$ -hybridized carbons.

Let's take the example of D-glucose (a D-aldohexose).

**$\beta$ -D-glucopyranose**



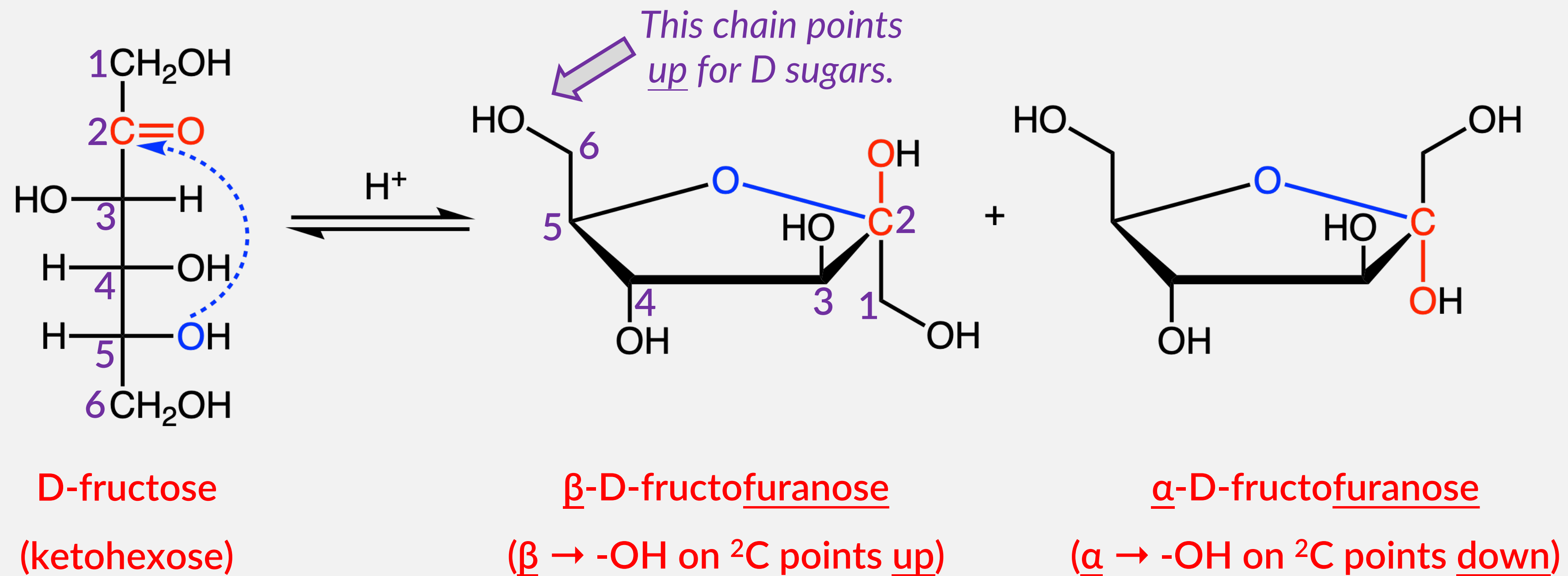
**$\alpha$ -D-glucopyranose**



# Haworth projections: Cyclic sugars (Ex. 2)

Sugars often cyclize into 5- (furanose) or 6-membered (pyranose) rings in of acidic ( $H^+$ ) medium.

Let's take the example of D-fructose (an D-ketohexose).



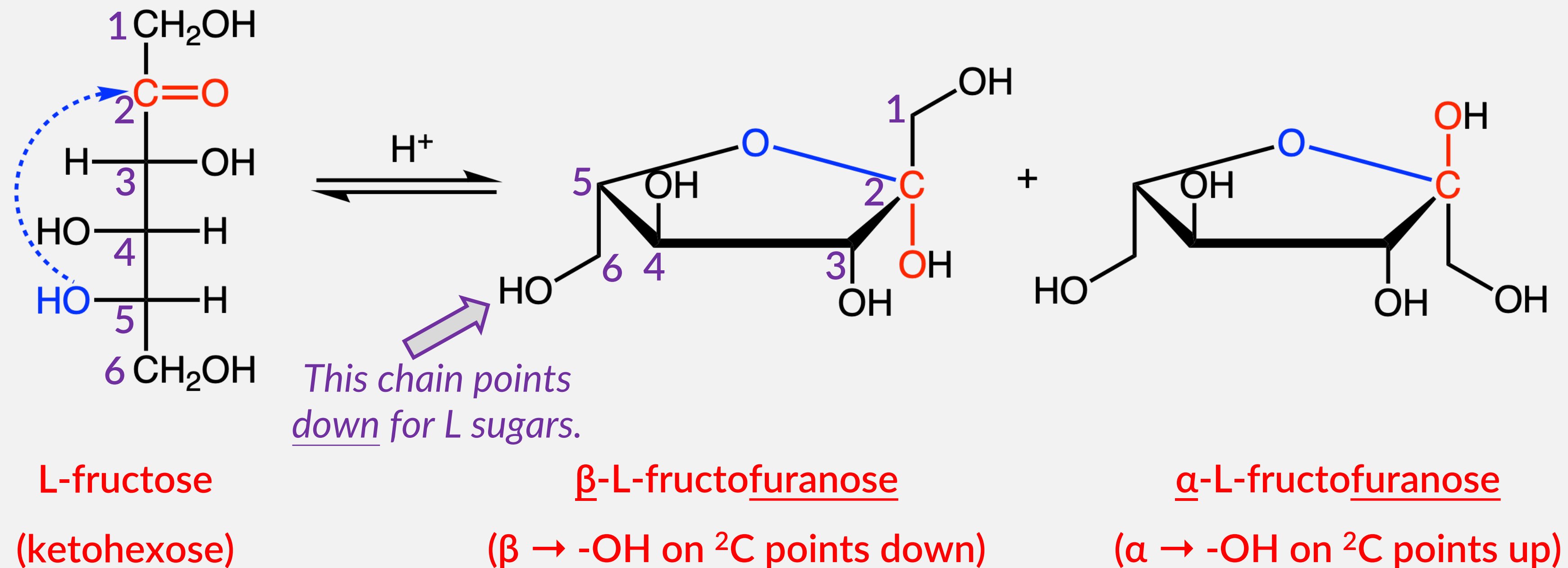
The anomeric carbon is the  $^2C$  which is the only carbon attached to two O atoms.

The  $\alpha$ - and  $\beta$ -furanoses are isomers that are called anomers.

# Haworth projections: Cyclic sugars (Ex. 3)

Most sugars are D-sugars. But what about L-sugars?

Let's take the example of L-fructose (an L-ketohexose).



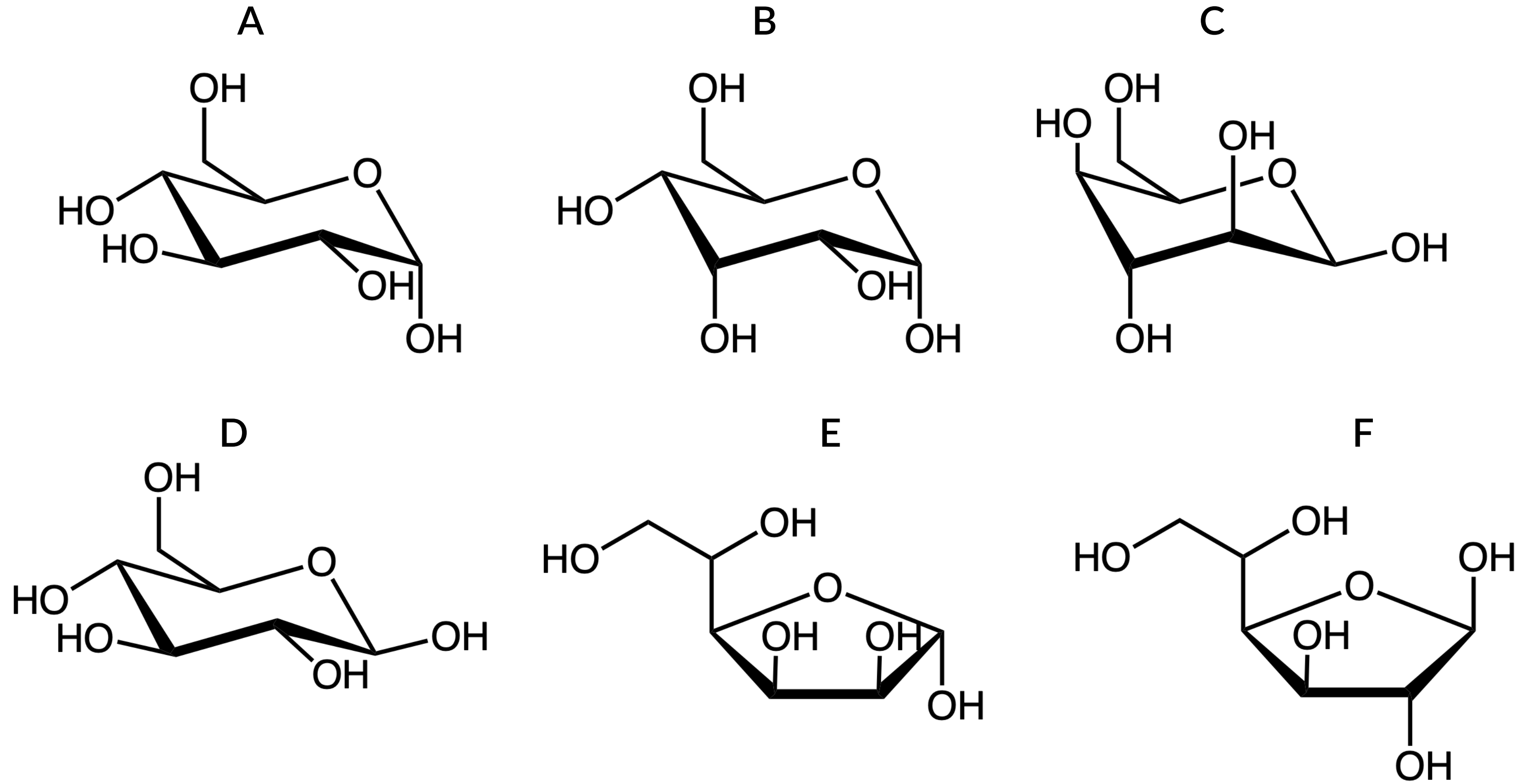
The anomeric carbon is the <sup>2</sup>C which is the only carbon attached to two O atoms.

The α- and β-furanoses are isomers that are called anomers.

# PRACTICE PROBLEM 1

Which of the following monosaccharides are  $\beta$ -anomers?

— answer —



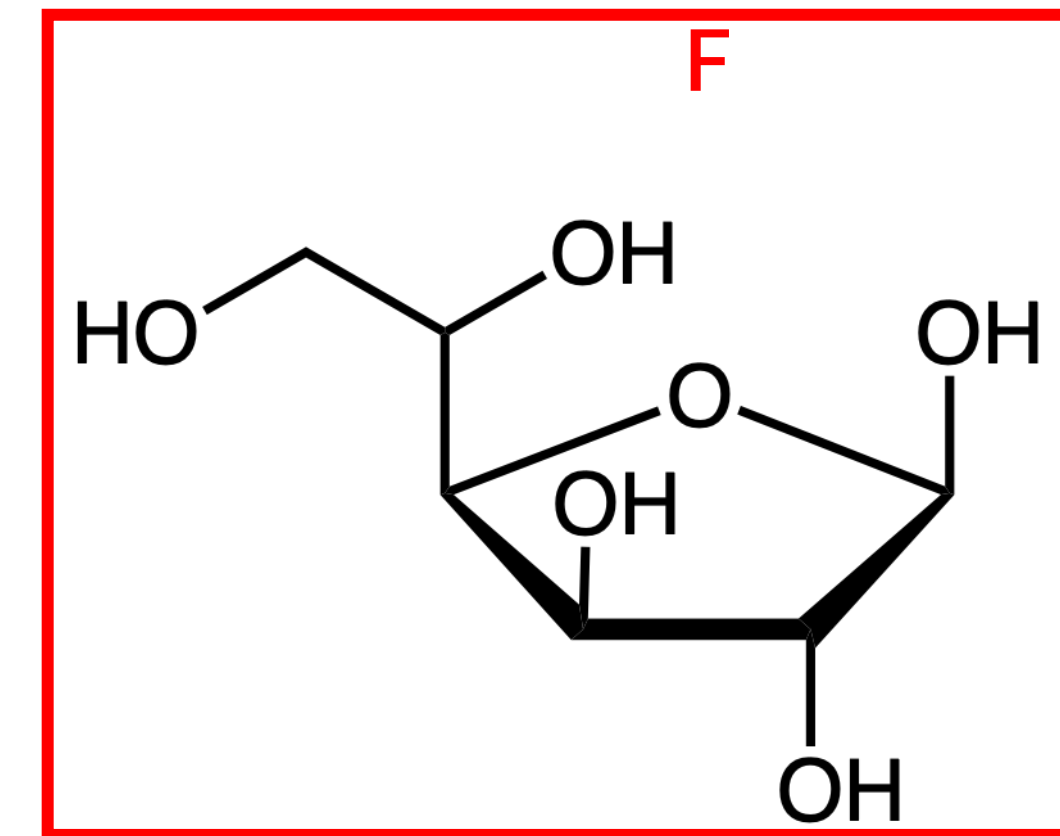
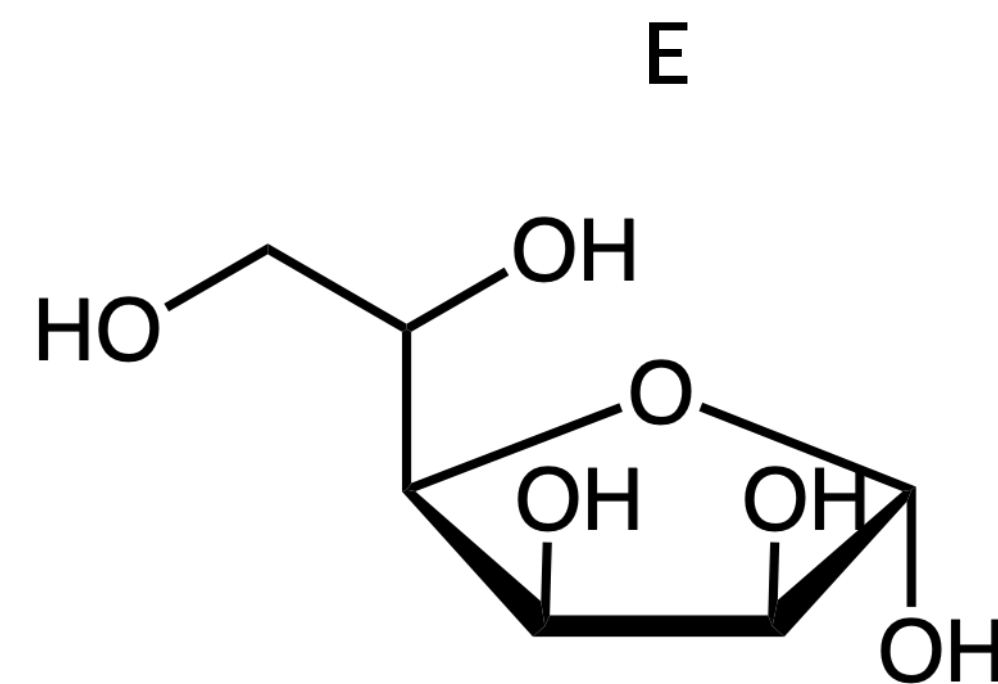
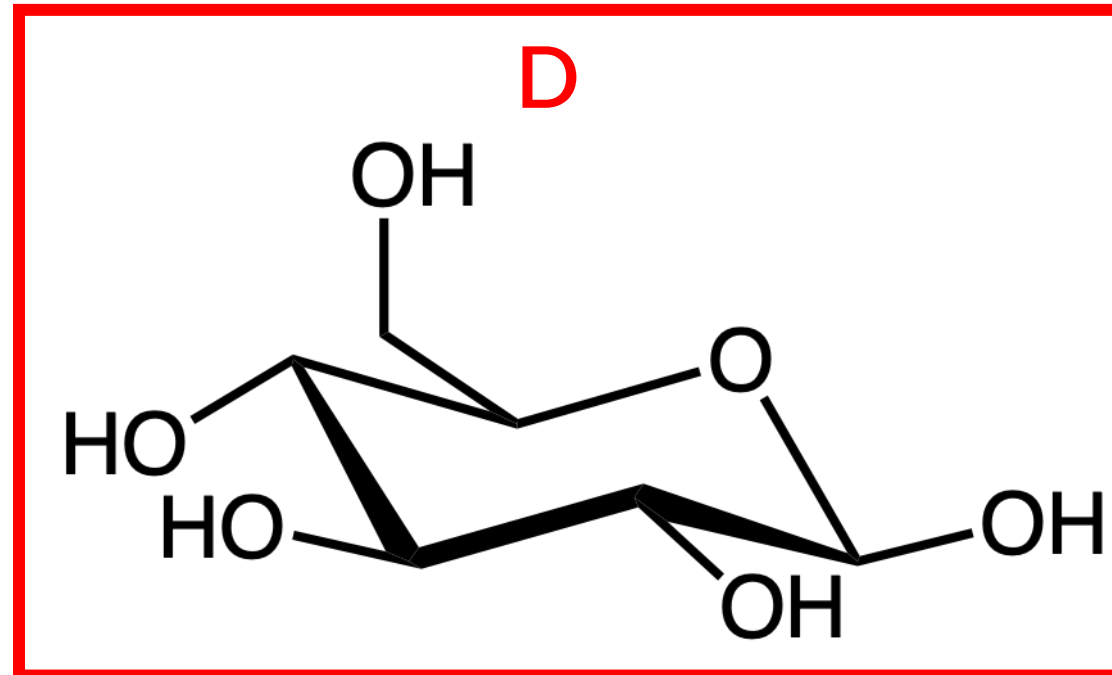
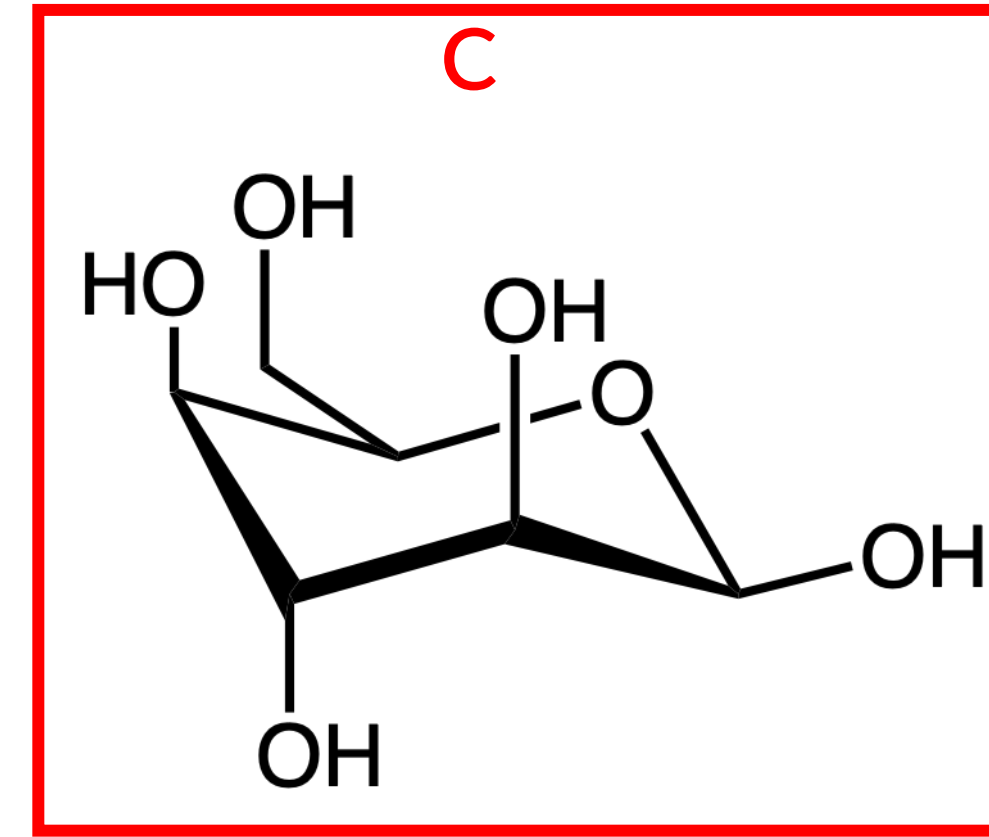
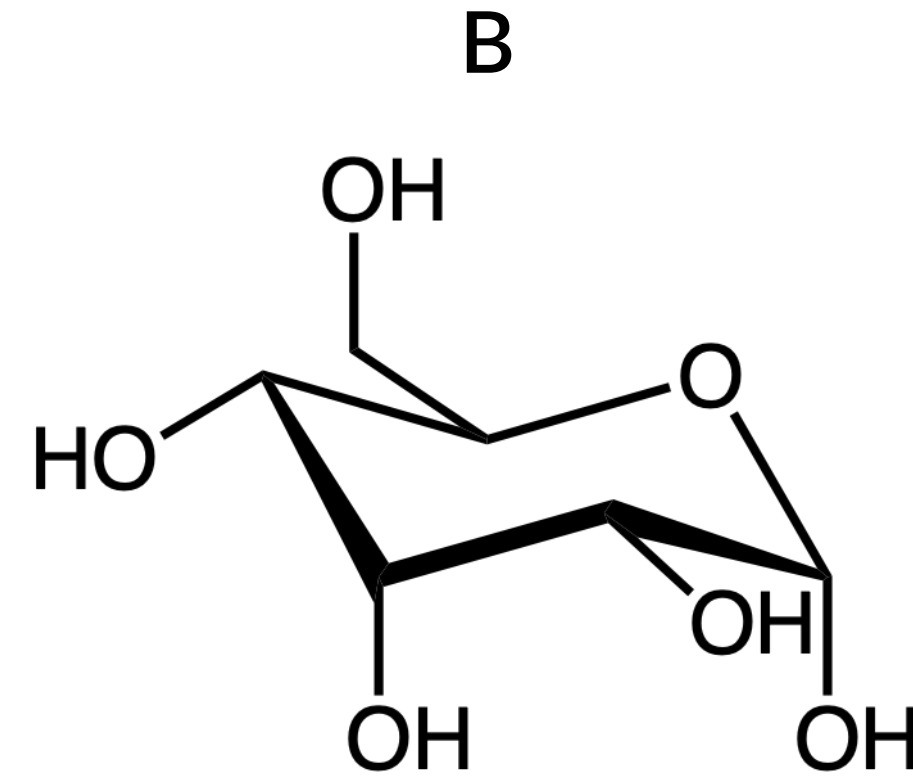
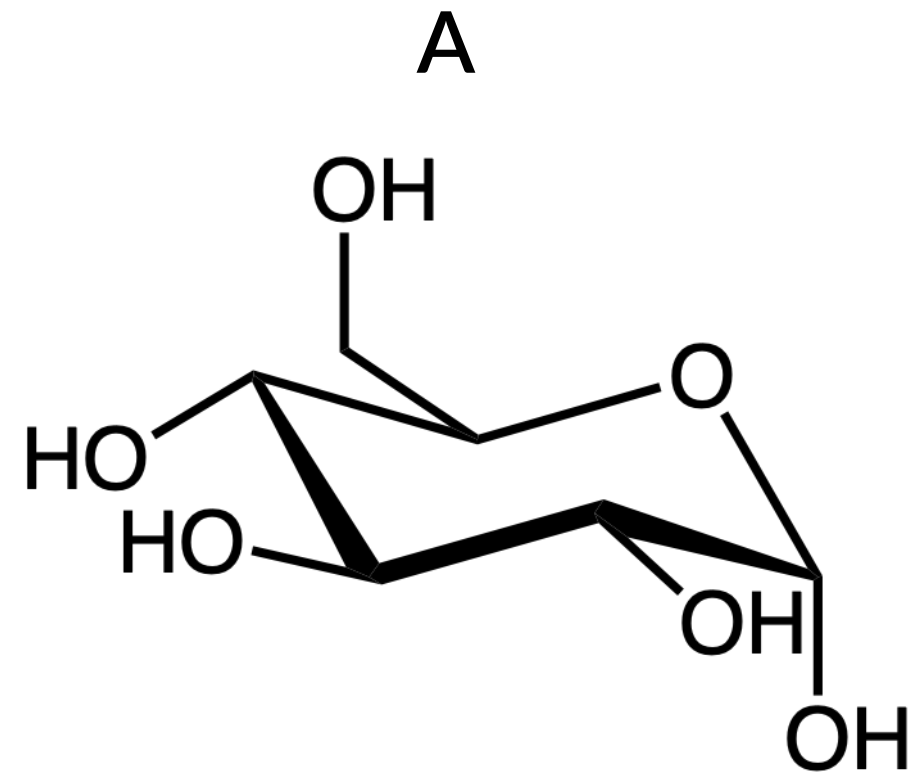


# PRACTICE PROBLEM 1

Which of the following monosaccharides are  $\beta$ -anomers?

— *answer* —

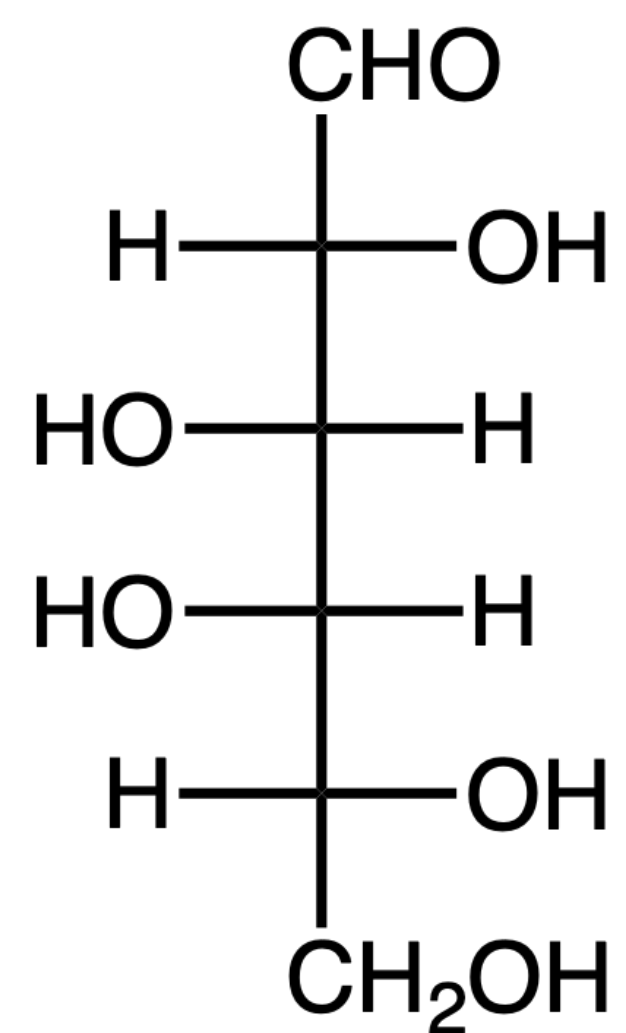
The  $\beta$ -anomer has the  $-OH$  on the anomeric carbon pointing up for these D-sugars.



## PRACTICE PROBLEM 2

Draw the Haworth projection for the  $\beta$ -pyranose form of D-galactose based on its Fischer projection.

— *answer* —



## PRACTICE PROBLEM 2

Draw the Haworth projection for the  $\beta$ -pyranose form of D-galactose based on its Fischer projection.

— *answer* —

The  $\beta$ -anomer has the  $\text{-OH}$  on the anomeric carbon pointing up for D-sugars.

